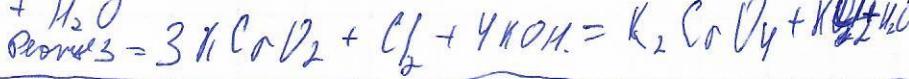
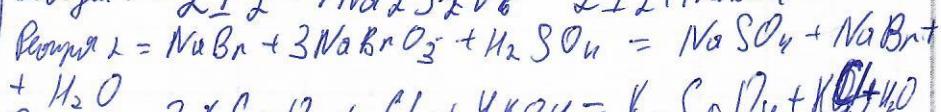
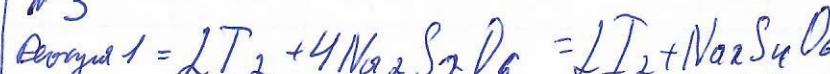


N1

$$1) \text{Na} = \frac{22,80}{24,64} \cdot 100\% = 92,04\%$$

$$2) V = \frac{25\% \cdot 1,175}{100\%} = 0,292$$

N3



N4

$$1) \frac{65,8 + 20 + 54,8 + 20}{100\%} = 139,8\%$$

$$2) m(\text{Cl}) = \frac{0,1412 + 0,632}{100\%} = 0,00853\text{ г}$$

3)

$$1) \text{N} = \text{Al}, \text{A} = \text{H}$$

$$2) \text{Al} + \text{H} = \text{HAl}$$

$$3) 20 : 1,25 = 16$$

$$1916 + 16 = 1932\text{ г}$$